

Nano-boric acid makes motor oil more slippery

One key to saving the environment, improving our economy and reducing our dependence on foreign oil might just be sitting in your mother's medicine cabinet.

Scientists at the U.S. Department of Energy's Argonne National Laboratory have begun to combine infinitesimal particles of boric acid – known primarily as a mild antiseptic and eye cleanser – with traditional motor oils in order to improve their lubricity and by doing so increase energy efficiency.

Ali Erdemir, senior scientist in Argonne's Energy Systems Division, has spent nearly 20 years investigating the lubricious properties of boric acid. In 1991, he received an R&D 100 award – widely considered the "Oscar of technology" – for showing that microscopic particles of boric acid could dramatically reduce friction between automobile engine parts. Metals covered with a boric acid film exhibited coefficients of friction lower than that of Teflon, making Erdemir's films the slickest solids in existence at that time.

"Ali was looking at large, micron-sized, particles," said George Fenske, who works alongside Erdemir at Argonne. "He was just sprinkling boric acid onto surfaces."

But driven by a conviction that he could fashion boric acid into an even better lubricant, Erdemir continued to chase the ultimate frontier: a perfectly frictionless material. Glimpsing the potential of nanotechnology, Erdemir went smaller – 10 times smaller – and was astonished by the behaviour of much thinner boric acid films. "If you can produce or manufacture boric acid at

the nanoscale, its properties become even more fantastic," he said.

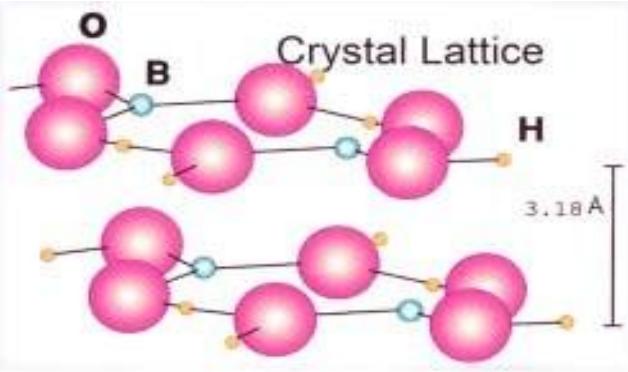
Reducing the size of the particles to as tiny as 50 nanometres in diameter – less than one-thousandth the width of a human hair – solved a number of old problems and opened up a number of new possibilities, Erdemir said. In previous tests, his team had combined the larger boric acid particles with pure poly-alpha olefin,



Argonne researcher Ali Erdemir performs a friction test on a metal disc coated with a solution of motor oil with nanoboric acid particles.

the principal ingredient in many synthetic motor oils. While these larger particles dramatically improved the lubricity of the pure oil, within a few weeks gravity had started to separate the mixture. By using smaller particles, Erdemir created a stable suspension of boric acid in the motor oil.

In laboratory tests, these new boric acid suspensions have reduced by as much as two-thirds the energy lost



This diagram illustrates the crystalline structure of boric acid. Boron atoms are shown as blue spheres, oxygen pink, and hydrogen brown. Molecular forces that bind the layers in the lattice enable them to slide over one another with very low friction.

through friction as heat. The implications for fuel economy are not hard to imagine, Erdemir said. "You're easily talking about a four or five percent reduction in fuel consumption," he said. "In a given day, we consume so many millions of barrels of oil, and if you can reduce that number by even one percent, that will have a huge economic impact."

Argonne is currently in talks with materials and lubricant manufacturers to bring boric acid technology to market, Erdemir said. While these new additives need to pass a battery of environmental and safety tests, they will probably be available within two years.

In his first experiments with boric acid, Erdemir demonstrated that the compound not only proved an effective lubricant but was also every industrial technologist's dream: It came from naturally abundant minerals, was cheap to manufacture, and posed no health hazards or environmental threats.

Boric acid owes its lubricious properties to its unique natural structure. The compound consists of a stack of crystallized layers in which the atoms tightly adhere to each other. However, these layers stack themselves relatively far apart, so that the intermolecular bonds – called van der Waals forces – are comparatively weak. When stressed, the compound's layers smear and slide over one another easily, like a strewn deck of playing cards. The strong bonding within each layer prevents direct contact between sliding parts, lowering friction and minimizing wear.

Until recently, most of Erdemir's work in boric acid lubrication had been restricted to motor oils, principally because of the relative bulk of the larger particles. The move to the nanoscale, however, has opened up other possible uses of the chemical. Through a simple chemical reaction, nano-boric acid can be transformed into a liquid relative of boric acid that has shown potential to increase fuel lubricity.

Using this liquid analog of solid boric acid as a fuel additive on a large scale could greatly benefit the environment, both because it would help to increase fuel efficiency and because it would replace existing fuel lubricants that are potentially harmful to the environment, Erdemir said. By themselves, most fuels – especially diesels – contain some sulphur and other special chemical additives to boost lubricity. When burned, however, some of these additives along with sulphur may cause harmful emissions and acid rain. However, the lack of a suitable alternative complicates efforts to cut sulphur content.

The substitution of liquid boric acid for sulphur-containing additives preserves the health of the car as well as that of the environment. Sulphur exhaust gradually coats the surface of a car's catalytic converter, the part that helps to reduce the toxicity of a car's emissions. Eventually, the converter becomes so choked with sulphur that it is no longer able to process any more exhaust.

Even though he has just begun to unleash the potential of boric acid, Erdemir believes that nanoscale synthetic compounds may prove to be even more effective lubricants. "The next step is to use the basic knowledge that we have gained out of this particular compound to come up with more exotic compounds that will work even better," he said.



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